LESSON PLAN

SUBJECT: APPLIED CHEMISTRY (TH 5(B))

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			CHAPTER	SUB CHAPTER
SI. No.	PERIOD S	TOPIC	CHAPTER	
1	PERIOD- 1	UNIT-1 (Atomic Structure, Chemical bonding and Solutions)	Atomic Structure	Introduction to Applied Chemistry Symbol, Valency and Chemical formulae of the compounds Atomic number & Atomic mass of element from Hydrogen to Zinc
2	PERIOD- 2	UNIT-1	Atomic Structure	Rutherford model of atom, Bohr's theory
3	PERIOD-	UNIT-1	Atomic Structure	Hydrogen spectrum explanation based on Bohr's model of atom, Heisenberg uncertainty principle
4	PERIOD-	UNIT-1	Atomic Structure	Quantum numbers – orbital concept
5	PERIOD-	UNIT-1	Atomic Structure	Orbital concept. Shapes of s,p and d orbitals,
6	PERIOD-	UNIT-1	Atomic Structure	Pauli's exclusion principle, Hund's rule of maximum multiplicity
7	PERIOD-	UNIT-1	Atomic Structure	Aufbau rule, electronic configuration.
8	PERIOD-	UNIT-1	Chemical Bonding	Concept of chemical bonding – cause of chemical bonding, types of bonds
9	PERIOD-	UNIT-1	Chemical Bonding	Ionic bonding (NaCl example), covalent bond (H2, F2)
10	PERIOD-	UNIT-1	Chemical Bonding	HF hybridization in BeCl2, BF3, CH4, NH3, H2O)
11	PERIOD- 11	UNIT-1	Chemical Bonding	Coordination bond in NH4+, and anomalous properties of NH3, H2O due to hydrogen bonding, and metallic bonding.
12	PERIOD- 12	UNIT-1	Solutions	Solution – idea of solute, solvent and solution, methods to express the concentration of solution molarity (<i>M</i> = mole per liter), ppm
13	PERIOD- 13	UNIT-1	Solutions	Mass percentage, volume percentage and mole fraction
14	PERIOD- 14	UNIT-1	Atomic Structure, Chemical bonding and Solutions	Revision, Assignment & Class test
15	PERIOD- 15	UNIT-2 (WATER)	WATER	Graphical presentation of water distribution on Earth (pie or bar diagram). Classification of soft and hard water based on soap test, salts causing water hardness

16	PERIOD-	UNIT-2	WATER	Unit of hardness and simple numerical on water
	16	(WATER)		hardness.
17	PERIOD- 17	UNIT-2 (WATER)	WATER	Cause of poor lathering of soap in hard water, problems caused by the use of hard water in boiler (scale and sludge)
18	PERIOD- 18	UNIT-2 (WATER)	WATER	problems caused by the use of hard water in boiler (foaming and priming, corrosion etc)
19	PERIOD- 19	UNIT-2 (WATER)	WATER	quantitative measurement of water hardness by ETDA method, Total dissolved solids (TDS) alkalinity estimation
20	PERIOD- 20	UNIT-2 (WATER)	WATER	Water softening techniques – soda lime process, zeolite process
21	PERIOD- 21	UNIT-2 (WATER)	WATER	Water softening techniques – Ion exchange process.
22	PERIOD- 22	UNIT-2 (WATER)	WATER	Municipal water treatment (in brief only) – sedimentation, coagulation
23	PERIOD- 23	UNIT-2 (WATER)	WATER	Municipal water treatment (in brief only)-filtration, sterilization
24	PERIOD- 24	UNIT-2 (WATER)	WATER	Water for human consumption for drinking and cooking purposes from any water sources and enlist Indian standard specification of drinking water (collect data and understand standards).
25	PERIOD- 25	UNIT-2 (WATER)	WATER	Revision, Assignment & Class test
26	PERIOD- 26	UNIT-3 (ENGINEERING MATERIALS)	ENGINEERING MATERIALS	Natural occurrence of metals – minerals, ores of iron, aluminium and copper
27	PERIOD- 27	UNIT-3	ENGINEERING MATERIALS	Gangue (matrix), flux, slag, metallurgy – brief acount of general principles of metallurgy.
28	PERIOD- 28	UNIT-3	ENGINEERING MATERIALS	Extraction of - iron from haematite ore using blast furnace
29	PERIOD- 29	UNIT-3	ENGINEERING MATERIALS	Aluminium from bauxite along with reactions
30	PERIOD- 30	UNIT-3	ENGINEERING MATERIALS	Alloys – definition, purposes of alloying, ferrous alloys and non-ferrous with suitable examples, properties and applications.
31	PERIOD- 31	UNIT-3	ENGINEERING MATERIALS	Port land cement and hardening
32	PERIOD- 32	UNIT-3	ENGINEERING MATERIALS	Glasses Refractory and Composite materials
33	PERIOD- 33	UNIT-3	ENGINEERING MATERIALS	Polymers – monomer, homo and co polymers, degree of polymerization, simple reactions involved in preparation
34	PERIOD- 34	UNIT-3	ENGINEERING MATERIALS	Application of thermoplastics and thermosetting plastics (using PVC, PS, PTFE, nylon – 6, nylon-6,6 and Bakelite)

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35	PERIOD- 35	UNIT-3	ENGINEERING MATERIALS	Rubber and vulcanization of rubber
16	PERIOD-	UNIT-3	ENGINEERING MATERIALS	Revision, Assignment & Class test
37	PERIOD- 37	UNIT-4 (CHEMISTRY OF FUELS &LUBRICANTS)	CHEMISTRY OF FUELS	Definition of fuel and combustion of fuel, classification of fuels, calorific values (HCV and LCV)
38	PERIOD-	UNIT-4	CHEMISTRY OF FUELS	Calculation of HCV and LCV using Dulong's formula.
39	PERIOD- 39	UNIT-4	CHEMISTRY OF FUELS	Proximate analysis of coal solid fuel petrol and diesel - fuel rating (octane and cetane numbers)
40	PERIOD- 40	UNIT-4	CHEMISTRY OF FUELS	Chemical composition, Calorific values and applications of LPG, CNG, water gas
41	PERIOD- 41	UNIT-4	CHEMISTRY OF FUELS	Chemical composition, Calorific values and applications of coal gas, producer gas and biogas
42	PERIOD-	UNIT-4	LUBRICANTS	Lubrication – function and characteristic properties of good lubricant
43	PERIOD-	UNIT-4	LUBRICANTS	classification with examples, lubrication mechanism – hydrodynamic and boundary lubrication
44	PERIOD-	UNIT-4	LUBRICANTS	physical proper- ties (viscosity and viscosity index, oiliness, flash and fire point)
45	PERIOD-	UNIT-4	LUBRICANTS	could and pour point, chemical properties (coke number)
46	PERIOD-	UNIT-4	LUBRICANTS	Total acid number Saponification value of lubricants.
47	PERIOD- 47	UNIT-5 (ELECTRO CHEMISTRY)	ELECTRO CHEMISTRY	Electronic concept of oxidation, reduction and redox reactions.
48	PERIOD- 48	UNIT-5	ELECTRO CHEMISTRY	Definition of terms: electrolytes, non-electrolytes with suitable examples, Faradays laws of electrolysis and simple numerical problems.
49	PERIOD- 49	UNIT-5	ELECTRO CHEMISTRY	Industrial Application of Electrolysis – Electrometallurgy, Electroplating, Electrolytic refining.
50	PERIOD- 50	UNIT-5	ELECTRO CHEMISTRY	Application of redox reactions in electrochemical cells – Primary cells – dry cell, Secondary cell - commercially used lead storage battery, fuel
51	PERIOD- 51	UNIT-5	ELECTRO CHEMISTRY	Solar cells. Introduction to Corrosion of metals – definition, types of corrosion (chemical and electrochemical)
52	PERIOD- 52	UNIT-5	ELECTRO CHEMISTRY	H ₂ liberation and O ₂ absorption mechanism of electrochemical corrosion, factors affecting rate of corrosion.





53	PERIOD- 53	UNIT-5	ELECTRO CHEMISTRY	Internal corrosion preventive measures –Purification, alloying and heat treatment
54	PERIOD- 54	UNIT-5	ELECTRO CHEMISTRY	External corrosion preventive measures: a) metal (anodic, cathodic) coatings,
55	PERIOD- 55	UNIT-5	ELECTRO CHEMISTRY	b) Organic Inhibitors.
56	PERIOD- 56	Revision Class	UNIT-1	Revision & Class Test
57	PERIOD- 57	Revision Class	UNIT-2	Revision & Class Test
58	PERIOD- 58	Revision Class	UNIT-3	Revision & Class Test
59	PERIOD- 59	Revision Class	UNIT-4	Revision & Class Test
60	PERIOD- 60	Revision Class	UNIT-5	Revision & Class Test

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