



# **UTKALMANI GOPABANDHU INSTITUTE OF ENGINEERING, ROURKELA**

## **LESSON PLAN**

**SESSION: 2023-2024**

**DEPARTMENT OF CERAMIC TECHNOLOGY**

**SUBJECT CODE: TH-2**

**NAME OF THE SUBJECT: CHEMISTRY OF CERAMIC MATERIAL**

**BRANCH: CERAMIC TECHNOLOGY**

**SEMESTER: DIPLOMA 3<sup>RD</sup> SEM**

**NUMBER OF CLASS ALLOTTED PER WEEK: 4(01-08-23 to 30-11-23)**

**TOTAL PERIODS ALLOTTED TO THE SUBJECT ACCORDING TO STEVT: 60**

**NAME OF THE FACULTY: KRUSHNA PRASAD DASH**

<b>Week/Date</b>	<b>Lecture</b>	<b>Topic to be covered</b>	<b>Remarks</b>
1 <sup>st</sup> week	1 <sup>st</sup>	Chapter-1: Thermodynamics and Thermo Chemistry 1.1: Define different terms of thermodynamic system.	
	2 <sup>nd</sup>	1.2: State ideal gas law.	
	3 <sup>rd</sup>	1.3: Explain Extensive and intensive properties.	
2 <sup>nd</sup> week	1 <sup>st</sup>	1.4: Define and explain Homogeneous and heterogeneous system.	
	2 <sup>nd</sup>	1.5: State and explain first law of thermo dynamics.	
	3 <sup>rd</sup>	1.6: Explain similarities between heat and work.	
	4 <sup>th</sup>	1.7: Define internal energy.	

3 <sup>rd</sup> week	1 <sup>st</sup>	1.8: Define heat capacity and specific heat.	
	2 <sup>nd</sup>	1.9: Define adiabatic Isothermal process.	
4 <sup>th</sup> week	1 <sup>st</sup>	1.10: Define heat of formation 1.11: define heat of reaction	
	2 <sup>nd</sup>	1.12: Limitation in the 1 <sup>st</sup> law of thermodynamics	
5 <sup>th</sup> week	1 <sup>st</sup>	1.13: explain the postulates of 2 <sup>nd</sup> law of thermodynamics	
	2 <sup>nd</sup>	1.14: Explain carnot cycle.	
	3 <sup>rd</sup>	1.15: Explain absolute scale of temperature.	
	4 <sup>th</sup>	1.16: State and explain free energy.	
6 <sup>th</sup> week	1 <sup>st</sup>	1.17: Define entropy and chemical potential.	
	2 <sup>nd</sup>	1.18: State the importance of thermodynamics in ceramics in ceramic industry.	
	3 <sup>rd</sup>	1.19: State and explain 3 <sup>rd</sup> law of thermo dynamics.	
	4 <sup>th</sup>	<b>CH. 2-PHASE RULE AND PHASE DIAGRAM</b> 2.1: Define phases, components, degree of freedom.	
7 <sup>th</sup> week	1 <sup>st</sup>	2.2: Phase diagram of pure substance Eutectic system, lever rule.	
	2 <sup>nd</sup>	2.3: Phase rule, one component, two component and three component system.	
8 <sup>th</sup> week		2.4: Explain importance of phase diagram and its application.	
		2.5: SiO <sub>2</sub> , Al <sub>2</sub> O <sub>3</sub> -SiO <sub>2</sub> , CaO-MgO and MgO-Al <sub>2</sub> O <sub>3</sub> system.	
		<b>Ch. 3- COLLOID</b> 3.1: Define and classify colloids.	
9 <sup>th</sup> week	1 <sup>ST</sup>	3.2: State & Explain different method of preparation of colloids.	

	2 <sup>ND</sup>	3.3:Application of colloids in ceramic making and in other areas.	
	3 <sup>RD</sup>	3.4:Base exchange capacity, flocculation, deflocculation & clay water system.	
10 <sup>th</sup> week	1 <sup>st</sup>	<b>Ch.4-THERMAL ANALYSIS AND SPECTROSCOPY.</b> 4.1:Introduction to thermo – gravimetric analysis.	
	2 <sup>nd</sup>	4.2:Basic idea on thermal method of analysis.	
	3 <sup>rd</sup>	4.3: State and Explain spectrophotometry concept in spectroscopy.	
11 <sup>th</sup> week	1 <sup>ST</sup>	4.4: Define spectroscopy and describe various types of spectrophotometers.	
	2 <sup>nd</sup>	4.6: Visible spectrophotometry and colorimetric.	
12 <sup>th</sup> week	1 <sup>ST</sup>	4.7: Discuss different application of spectrophotometric.	
	2 <sup>ND</sup>	4.8: Description of atomic absorption spectrophotometers.	
	3 <sup>RD</sup>	4.9: Introduction DTA, TGA, XRD, & XRF.	

